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(54) Title: POSITIONABLE ADHESIVE SYSTEM WITH HIGH SHEAR STRENGTH

(57) Abstract

A positionable aqueous adhesive composition is provided, comprising a blend or mixture of (a) an aqueous suspension of hollow, polymeric, inherently tacky elastomeric microspheres, and (b) an aqueous film-forming pressure sensitive adhesive latex. The adhesive is positionable, in that it allows for positional adjustment after application to a receptor, yet displays high peel adhesion and shear strength.

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- 1 -

POSITIONABLE ADHESIVE SYSTEM WITH HIGH SHEAR STRENGTH

Field of the Invention

This invention relates to a positionable adhesive system displaying final elevated shear strength and peel adhesion, comprising a mixture of hollow, polymeric, acrylate, infusible, inherently tacky elastomeric microspheres together with a polymeric film-forming pressure sensitive latex adhesive.

Background of the Invention

Description of the Related Art

Positionable adhesives are those which allow for placement of an article containing such an adhesive onto a receptor in an exact position, because the article can be adjusted relative to the receptor after initial placement.

In some instances, the adhesive can be designated repositionable or repeatedly reuseable. Such adhesives exhibit aggressive tack but low peel adhesion properties, thus allowing repeated reuseability. Commercial products such as the 3M Brand Post-itTM Notes display such adhesive characteristics.

Positionable adhesive systems, however, as used herein, generally display high peel adhesion and elevated shear strength after positioned placement.

Such systems have been disclosed for utility such as in the creation of sign faces or other graphic systems utilizing a vinyl film. For example, U.S. Patent Nos. 3,314,838; 3,331,729 and 3,413,168 disclose the concept of positionability based on a conventional film-forming adhesive having contained therein frangible glass bubbles. Positionability is obtained by the fact that the glass bubbles, during initial application of the coated article,

- 2 -

allow for incomplete surface contact of the adhesive to the ultimate receptor. Once the article is properly positioned, pressure on the face of the article causes a rupture of the glass bubbles, thus allowing the full 5 surface area to be contacted with the receptor, forming a strong bond. Obviously, once the frangible bubbles are ruptured, any further positionability characteristics are lost. In addition, articles coated with the adhesive containing the frangible bubbles 10 inherently have added manufacturing steps and furthermore require special liners for application.

Reported adhesive systems for repositionable products range from the microsphere adhesives disclosed in U.S. Patent Nos. 3,691,140 and 4,166,152 to 15 pattern-coated conventional adhesive systems. A further definition of an adhesive system containing such microspheres is disclosed in U.S. Patent No. 3,857,731, wherein a binder material is mixed with the microspheres. In some examples in this latter patent, 20 the binder is an acrylate pressure sensitive adhesive. However, in all instances, the adhesion characteristics of the microspheres themselves are not taught to be overshadowed by the pressure sensitive adhesive. In this instance, the binder creates a physical "socket" 25 to mechanically hold the individual microspheres onto a substrate, thus asserting to prevent undesired transfer of the microspheres to a receptor, increasing the retention of desired adhesive features.

U.S. Patent No. 4,735,837 discloses a similar 30 system, wherein solid microspheres are mixed with a binder resin, which is taught to be a pressure sensitive adhesive. In this disclosure, the characteristics of adhesion and repositionability come from the pressure sensitive adhesive rather than the 35 microspheres themselves. In fact, the patent discloses that the microspheres need not have adhesion capabilities at all.

- 3 -

In EPO 209,337, it is disclosed that the basic deficiency of solid microsphere adhesives is that they exhibit excessive transfer. In this instance, a reactive monomer is included with the adhesive 5 microspheres during their preparation. The monomer remains unreacted during the adhesive polymerization, but subsequently undergoes a reaction with the binder and/or the backing to provide shear values of up to 50 Newtons, claimed to be higher than other 10 microsphere-based systems.

In contrast with the foregoing, we have now discovered that an aqueous pressure sensitive adhesive system can be made utilizing a combination of hollow microspheres which are polymeric, acrylate, infusible 15 and inherently tacky, together with a polymeric film-forming pressure sensitive adhesive. This aqueous blend, when applied to a backing, displays the desired positionability, ease of application, and typically associated with the positionable adhesives discussed 20 above, and in addition displays the high shear strength, high peel adhesion and other performance characteristics normally associated with conventional film-forming pressure sensitive adhesives. In addition, we have found that this system provides for 25 clean removal from most receptors.

Summary of the Invention

In accordance with the invention there is 30 provided a positionable aqueous adhesive composition and article containing same thereon, the composition comprising a blend or mixture of (a) an aqueous suspension of hollow, polymeric, infusible, inherently tacky elastomeric microspheres, and (b) an aqueous 35 film-forming pressure sensitive adhesive latex comprising at least one long chain alkyl acrylate. Preferably, the weight ratio of microspheres to latex,

- 4 -

based on solids, is from about 12 to 1 to about 39 to
1.

The adhesive is positionable, i.e., can be
adjusted positionally after application to a receptor,
5 yet displays high peel adhesion and shear strength, and
displays clean removal from most receptors.

Graphic or decorative films, e.g., polyvinyl
chloride can be applied to a receptor and positioned
accurately, yet final peel adhesion and shear forces
10 will be sufficiently elevated to allow the film to
remain in place without movement or deformation.

Detailed Description of the Invention

The hollow polymeric microspheres having
15 utility herein are disclosed in detail in commonly
assigned copending application Serial No. 276,767,
incorporated herein by reference. Such microspheres
comprise at least about 85 parts by weight of at least
one alkyl acrylate or alkyl methacrylate ester together
20 with up to about 15 parts by weight of at least one
polar monomer, a majority of such microspheres having
one or more interior voids. Preferably, the
microspheres contain a central cavity at least 10%, and
most preferably, 30% of the diameter of the microsphere
25 itself. An aqueous suspension of such hollow
microspheres can be prepared by emulsification
processes as disclosed in the subject application.
Basically, they can be prepared by a two-step process
comprised of forming a water-in-oil emulsion of an
30 aqueous solution of one or more polar monomers in oil
phase monomers; forming a water-in-oil-in-water
emulsion by dispersing the first emulsion into an
aqueous phase; followed by initiating polymerization,
preferably by the application of heat or radiation. In
35 addition, aqueous suspensions of such microspheres
which contain moderately ionized polar monomers may be
prepared by a simpler one-step process comprising the
aqueous suspension polymerization of at least one alkyl

- 5 -

acrylate or methacrylate ester monomer and at least one non-ionic polar monomer in the presence of at least one emulsifier capable of producing a water-in-oil emulsion inside the droplets, which is substantially stable

5 during polymerization. Both methods produce an aqueous suspension of monomer droplets which upon polymerization become microspheres, the majority of which contain at least one interior cavity as discussed above.

10 The aqueous pressure sensitive adhesive latex used to blend with the aqueous suspension of hollow microspheres is composed of at least one long chain alkyl acrylate, i.e., containing from 4 to 12 carbon atoms, and preferably a polar monomer.

15 The film-forming pressure sensitive adhesive should be used at a level lower than the interspatial volume of the hollow microspheres. Preferably, the adhesive mixture contains a microsphere to film-forming adhesive ratio of from 39:1 to 12:1. Levels of
20 film-forming adhesive significantly higher than this may produce adhesive coated articles that retain the characteristics of the film-forming pressure sensitive adhesive, detrimental to positionability.

The aqueous composition can be conventionally applied onto materials such as paper, and vinyl and polyester films, as by knife or notched-bar coating techniques, for example.

30 Additives can be included in the mixture to enhance specific objectives, such as the coating of the aqueous system or modifying performance characteristics thereof.

Anti-foams can be added, especially when the aqueous system of the invention is to be applied as a coating at high rates of speed. One example of such a
35 material is "Foamaster" JMY, a conventional anti-foam agent from Henkel Process Chemicals, Inc.

- 6 -

Colorants can be added to enhance visual appearance, quality control, etc. Several vendors provide concentrates which add such desirable features.

Chemical crosslinkers, such as polyaziridines, 5 can increase ply adhesion, high temperature shear and cohesive strengths. Such improvements, however, typically can provide a loss in tack and ultimate adhesion.

Viscosity control materials can be added to 10 assist in coating quality and control. Examples include water soluble materials such as various cellulose products, polyacrylic acids, and vinyl alcohols.

The invention will now be described by the use 15 of the following non-limiting examples, wherein all parts are by weight unless otherwise specified.

Example 1

Hollow microspheres are manufactured in 20 accordance with the following procedure:

In a one-liter resin reactor equipped with mechanical stirrer, thermometer and inlet-outlet lines for vacuum and nitrogen, 450 grams of deionized water, 141 grams of iso-octyl acrylate, 9.0 grams of acrylic acid and 0.5 gram of benzoyl peroxide were charged.

Vacuum was applied to evacuate the reactor atmosphere and the reactor was purged with argon. The agitation was set to 400 RPM and when the initiator had dissolved, 1.5 grams of ammonium lauryl sulfate 30 (Standapol A, from Henkel AG) were added. The temperature of the reactor was raised to 60°C and maintained for 22 hours, an argon purge being maintained during the polymerization. The suspension was then allowed to cool to room temperature. The 35 reactor was then emptied and the suspension filtered. Optical microscopy showed hollow microspheres from about 4 to about 90 microns in diameter, the majority

- 7 -

of the microspheres containing a central cavity of at least 30% of the diameter of the microspheres.

This microsphere suspension exhibits a pH typically between 2.0 and 4.0 and has a solids content 5 of approximately 25%. With this procedure, the mean sphere diameter is typically 40-60 microns. (Different vessel proportions or mixing rates will alter the particle size, void diameter, etc.)

An aqueous pressure sensitive adhesive was 10 prepared as follows:

A split-resin flask of 2000 ml. capacity was fitted with a variable speed agitator, condenser, purging tube for introducing nitrogen, and a recording controller. The following materials were added to the 15 flask while purging same with nitrogen:

	404 grams	Distilled water
	1.50 grams	Sodium dodecylbenzene sulfonate
	435 grams	Isooctyl acrylate
20	60 grams	N-tert-octylacrylamide
	0.60 gram	Sodium bicarbonate
	5.0 grams	Sodium styrene sulfonate

The solid N-tert-octylacrylamide was dissolved in the 25 isooctyl acrylate before adding to the flask. The nitrogen purge was continued until the end of the run. The flask and its contents were heated to 50°C, at which temperature an initiator charge of 0.05 gram potassium persulfate and 0.0125 gram sodium 30 metabisulfite was added. The reaction temperature was held at 50°C for approximately 24 hours to complete the polymerization. The resulting latex had no coagulum and the solids content was 54%.

The adhesives as prepared above were combined 35 by adding, to a suitable container, 95.88 parts of the aqueous microsphere suspension at 25% solids; 3.6 parts of the aqueous pressure sensitive adhesive containing 54% solids; 0.52 part of ASE-60, commercially available

- 8 -

from Rohm & Haas, as a rheological control agent. The pH was then raised with ammonium hydroxide to between 7.5 and 8.5.

The composition was then knife coated onto a 5 polyvinyl chloride film and dried at 200°F for 1 minute to provide a dry coating weight of 19 grams per square meter.

Example 2

The adhesive of Example 1 was duplicated with 10 the exception of utilizing 98.77 parts of the microsphere suspension and 1.23 parts of the pressure sensitive adhesive latex. This was applied to a polyvinyl chloride film as per Example 1.

15

Example 3

An adhesive blend was prepared by mixing, in a suitable container, 96.23 parts of the aqueous microsphere suspension of Example 1; 2.99 parts of Gelva 2397, an aqueous pressure sensitive adhesive 20 available from Monsanto, containing 65% solids; 0.52 part of ASE-60; followed by raising the pH with ammonium hydroxide to between 7.5 and 8.5; and the addition of 0.26 part of C-E2 color concentrate from Ciba-Geigy to provide opacity.

25

The composition was coated onto a polyvinyl chloride film at a dry coating weight of 19 grams per square meter, as per Example 1.

Example 4

30 A film-forming pressure sensitive adhesive was manufactured as follows:

A split-resin flask of 2000 ml. capacity was fitted with a variable speed agitator, condenser, purging tube for introducing nitrogen, and a recording 35 controller. The following materials were added to the flask while purging the flask with nitrogen:

600 grams Distilled water

- 9 -

	4.80 grams	Sodium dodecylbenzene sulfonate
	4.80 grams	Nonyl phenol - 10.5 mole ethylene oxide adduct
	16.0 grams	Acrylic acid
5	160.0 grams	Ethyl acrylate
	160.0 grams	Isooctyl acrylate
	64.0 grams	Butyl acrylate

The nitrogen purge was continued until the end of the run. The flask and its contents were agitated at 300 rpm, and was heated to 32°C. At this temperature an initiator charge of 0.30 gram potassium persulfate and 0.08 gram sodium metabisulfite was added. An exothermic reaction occurred increasing the temperature to approximately 75°C after which the reaction mixture was allowed to cool. The resulting latex had no coagulum and the solids content was 40%.

An adhesive blend was then made by mixing 97.68 parts of the aqueous microsphere suspension of Example 1 with 1.57 parts of the film-forming pressure sensitive adhesive, together with 0.5 part of ASE-60, adjustment of pH to 7.5 to 8.5 with ammonium hydroxide, and the addition of 0.25 part of C-E2 color concentrate.

25 The blend was coated as per Example 1 onto a polyvinyl chloride film at a coating weight of 19 grams per square meter.

Example 5

30 A film-forming pressure sensitive adhesive was made in accordance with Example 4 with the exception that the monomer mixture was comprised of 320 grams of ethyl acrylate, 64 grams of butyl acrylate, and 16 grams of acrylic acid. A blend was then prepared by adding 94.46 parts of the aqueous microsphere suspension of Example 1 to 4.77 parts of the film-forming pressure sensitive adhesive, again containing 40% solids, 0.52 part of ASE-60, pH

- 10 -

adjustment to 7.5 to 8.5 with ammonium hydroxide, and the addition of 0.26 part of C-E2 color concentrate.

Again, the blend was applied onto polyvinyl chloride film at a coating weight of 19 grams per 5 square centimeter as per Example 1.

The foregoing samples were tested for adhesion, both initially and after aging, and dynamic shear was also determined. These results are reported in TABLE 1.

10 For the adhesion tests, a 25.4 mm strip of sample was applied to a test panel using three passes with a 2.27 kg roller. The adhesion was tested by peeling the adhered sample at 180 degrees at a rate of 2.57 mm per minute. The value reported is an average 15 reading over a five second period.

Samples were also aged at room temperature (27°C) for the designated time.

For the dynamic shear test, a .076 mm polyester tape was laminated to a 26.5 mm wide sample to be 20 tested. The samples were adhered to a test panel, typically a painted metal panel, and trimmed such that a 25.7 mm by 25.7 mm contact area was retained. The sample was pulled with an Instron testing machine at a rate of 5 mm/minute. The maximum force required to 25 remove the sample from the test panel is recorded.

TABLE 1

	Initial Adhesion, Newtons Per Decimeter	Adhesion		
		After 24 Hours,	Dynamic shear, Newtons per Decimeter	Newtons Per Decimeter <u>Width</u>
35	Example 1	20.84	27.50	1176.92
	Example 2	22.07	20.97	1117.99

- 11 -

Example 3	23.67	26.45	1228.06
Example 4	20.67	26.09	993.18
5 Example 5	27.34	27.32	1037.52

Visual observation of the samples showed no sign of adhesive transfer to the receptor test
10 specimens.

For best performance, the blend should be applied at the minimum coating thickness required to obtain a monolayer of microspheres. High coating thicknesses are basically wasteful, while reduced
15 coating thicknesses can result in incomplete surface coverage, and thus reduced peel adhesion and shear values. The preferred coating thickness is represented by the corresponding diameter of the average microsphere. For example, microspheres made by the
20 aforementioned procedure typically average 40 to 60 microns in diameter, and thus the preferred coating thickness would be in that same range.

- 12 -

In the Claims

1. An aqueous adhesive composition capable of displaying positionability, when applied as a layer to a
5 substrate, while maintaining elevated shear strength and peel adhesion, comprising an aqueous blend of
(a) an aqueous suspension of hollow, polymeric, infusible, inherently tacky elastomeric microspheres; and
10 (b) an aqueous film-forming pressure sensitive adhesive latex comprising at least one long chain alkyl acrylate.

2. The composition of claim 1 wherein said
15 microspheres are comprised of at least about 85 parts by weight of at least one alkyl acrylate or methacrylate ester and up to about 15 parts by weight of at least one polar monomer.

20 3. The composition of claim 1 wherein the weight ratio of said microspheres to said latex, on a solids basis, is from about 12 to 1 to about 39 to 1.

4. The composition of claim 1 wherein said
25 latex further comprises at least one polar monomer.

5. An article comprising a substrate, having applied to at least a portion of at least one surface thereof, a coating of a positionable adhesive
30 composition comprising
(a) an aqueous suspension of hollow, polymeric, infusible, inherently tacky elastomeric microspheres; and
(b) an aqueous film-forming pressure sensitive adhesive latex comprising at least one long chain alkyl acrylate.
35

6. The article of claim 5 wherein said substrate is a polyvinyl chloride film.

- 13 -

7. The article of claim 5 wherein said microspheres are comprised of at least about 85 parts by weight of at least one alkyl acrylate or methacrylate ester and up to about 15 parts by weight 5 of at least one polar monomer.

8. The article of claim 5 wherein the weight ratio of said microspheres to said latex, on a solids basis, is from about 12 to 1 to about 39 to 1.

10

9. The article of claim 5 wherein said latex further comprises at least one polar monomer.

INTERNATIONAL SEARCH REPORT

International Application No.

PCT/US 92/00613

I. CLASSIFICATION OF SUBJECT MATTER (If several classification symbols apply, indicate all)⁶

According to International Patent Classification (IPC) or to both National Classification and IPC

Int.Cl. 5 C09J7/02; C09J133/08

II. FIELDS SEARCHED

Minimum Documentation Searched⁷

Classification System	Classification Symbols
Int.Cl. 5	C09J

Documentation Searched other than Minimum Documentation
to the Extent that such Documents are Included in the Fields Searched⁸III. DOCUMENTS CONSIDERED TO BE RELEVANT⁹

Category ¹⁰	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³
Y	US,A,4 968 562 (J.DELGADO) 6 November 1990 see column 5, line 44 - line 49 see column 8, line 1 - line 7; claims 1-11 ----	1-5
Y,P	EP,A,0 419 020 (3M) see claims 1-4,6,9,10 ----	1-5

¹⁰ Special categories of cited documents:

- "A" document defining the general state of the art which is not considered to be of particular relevance
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IV. CERTIFICATION

Date of the Actual Completion of the International Search

1

15 JUNE 1992

Date of Mailing of this International Search Report

10.07.92

International Searching Authority

EUROPEAN PATENT OFFICE

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ANNEX TO THE INTERNATIONAL SEARCH REPORT
ON INTERNATIONAL PATENT APPLICATION NO. US 9200613
SA 57768

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Patent document cited in search report	Publication date	Patent family member(s)		Publication date
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